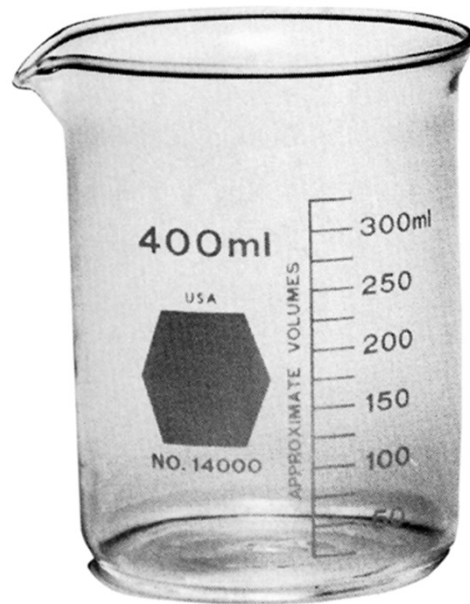


# The Naked Egg

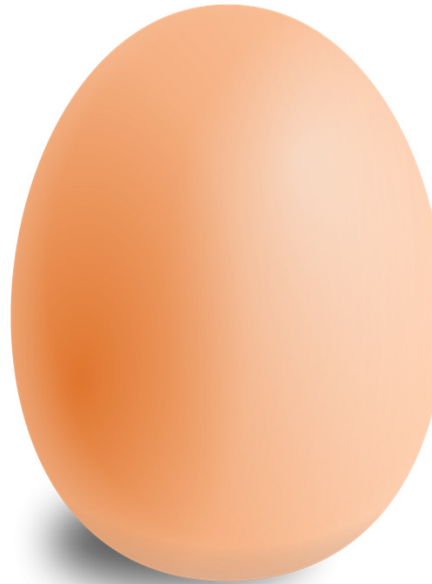


# We need...

1. A beaker



2. An egg



3. Some vinegar



## Procedure:

Place the egg into the beaker and then pour in vinegar until the egg is covered.

# What happened?

Bubbles came off of the egg when it was covered with vinegar. The outer shell contains 'calcium carbonate' and the vinegar reacts with the egg shell. When this happens bubbles (carbon dioxide) are made. This continues until all of the egg shell has reacted with the vinegar.

The inner egg shell is left and that is softer than the outer shell.